ayny

Stereoselective Preparation of Spiro[4.4] Cyclic Compounds by the Photochemical Activation of Oxazoles

Noriyoshi Arai,* Moe Mizota, and Takeshi Ohkuma*

Division of Chemi[cal](#page-2-0) Process Engineering and Frontier Chemistr[y C](#page-2-0)enter, Faculty of Engineering, Hokkaido University, Sapporo, Hokkaido 060-8628, Japan

S Supporting Information

[AB](#page-2-0)STRACT: [A novel pho](#page-2-0)tocyclization of 2-aryloxazole derivatives linked with an alkene moiety through a three-atom spacer at the 5 position gave a range of functionalized spiro $[4.4]$ cyclic compounds that included cyclopentane, tetrahydrofuran, and pyrrolidine moieties in moderate to high yields with excellent diastereoselectivity.

Photochemical reactions have long been utilized in organic synthesis as a powerful tool for realizing molecular complexity.^{1,2} Photochemical transformation can be performed by simply irradiating a solution containing a starting material to provide a [met](#page-3-0)al-free access to complex molecules from readily available compounds. The $[2 + 2]$ cycloaddition reaction,³ the meta-photocycloaddition onto arenes,⁴ and the di- π -methane rearrangement⁵ are [t](#page-3-0)he typical examples employed in total syntheses of natural products. In c[on](#page-3-0)trast to these widely employed re[ac](#page-3-0)tions, the photochemical reaction of fivemembered heteroaromatic compounds was less explored,^{6−11} though they are a common motif in organic chemistry. We focused our attention on the unique photochemical behavi[or of](#page-3-0) five-membered heteroaromatics to develop new and efficient chemical transformations, and identified several interesting photocyclizations of furans and thiophenes.^{12,13} In the course of our investigation, we found that oxazoles show quite interesting behavior under photochemical condition[s to](#page-3-0) give rise to an unprecedented intramolecular cyclization. Bond formation at the 4- or 5-position of oxazoles accompanied by dearomatization provides an efficient approach to the construction of a protected 1,2-aminoalcohol moiety.^{9,14} We report herein the novel photocyclization of 2-aryloxazoles giving functionalized spiro[4.4] cyclic compounds inclu[ding](#page-3-0) cyclopentane, pyrrolidine, and tetrahydrofuran skeletons in a highly stereoselective manner.

We synthesized 2-phenyloxazole derivative 1a, which had an alkene moiety, through a three-carbon spacer at the 5-position for the substrate of the photoreaction. External irradiation of a solution of 1a in benzene by a high-pressure mercury lamp through Pyrex glass afforded an unexpected cyclized product 2a in moderate yield (estimated from ¹H NMR integrals) with concomitant formation of the $[2 + 2]$ addition product 3 (Scheme 1). It was quite surprising that the unusual spirocyclic compound 2a was preferably produced over the conventional $\begin{bmatrix} 2 & + & 2 \end{bmatrix}$ adduct $\overline{3}^{9a-d,15}$ More importantly, the reaction proceeded in a highly stereoselective manner to give 2a as a Scheme 1. Stereoselective Functionalized Cyclopentane Formation from an Oxazole Derivative 1a

single diastereomer with a cis-configuration of the alkenyl group and the alkoxy moiety. The structure of 2a was deduced by careful spectroscopic analyses, including NMR $(^1H,~^{13}C,$ 2Ds, and NOE), HRMS, and IR (see the Supporting Information). We were interested in this new photochemical transformation of oxazole derivatives and decided t[o investigate](#page-2-0) [the reaction](#page-2-0) in detail.

At the beginning of our study, we tried the reaction in a series of solvents. The results are summarized in Table 1. The reaction proceeded in moderately polar solvents such as 1,4 dioxane, MTBE,¹⁶ and ethyl acetate as well as in b[en](#page-1-0)zene (entries 1−4), while the reaction in hexane, acetonitrile, and methanol gave a [lo](#page-3-0)wer yield of 2a with 8−20% recovery of the starting material 1a (entries 5−7). The production of 2a was completely inhibited in acetone (entry 8). In benzotrifluoride, 17 although 1a was consumed completely, 2a was obtained in only 38% yield with formation of an uncharacterizable mixt[ure](#page-3-0) (entry 9).

Setting benzene as a suitable solvent for this reaction, we carried out the reaction using a range of oxazole derivatives. The results are shown in Table 2. The reactions shown in Table

Received: November 12, 2014

Published: December 17, 2014

Table 1. Solvent Screen^a

^aAll reactions were carried out in a Pyrex test tube by external irradiation at a concentration of $2.7-2.9$ mM. ^bDetermined by ¹H NMR integral ratio using 1,1,2,2-tetrachloroethane as an internal standard. "Methyl *tert*-butyl ether.

2 were conducted in a photochemical reaction vessel for the internal irradiation. A yield of 2a even higher than that by external irradiation was achieved (Table 2, entry 1 vs Table 1, entry 1). Although 2a was somewhat susceptible to hydrolytic decomposition during the conventional chromatographic purification, we could isolate the compound in pure form with only a slight loss of yield. The substituents at the alkene moiety had a notable impact on the reaction. The reaction of 1b with a tetrasubstituted alkenyl group revealed that the bulkiness of the alkene did not affect the reaction (entry 2). It should be noted that neighboring quaternary carbon centers were created stereoselectively, suggesting the high potential of this reaction for the synthesis of organic molecules with a congested structure. Irradiation of the substrate 1c, which has an (E)-crotyl side chain instead of prenyl, gave vinyl-substituted cyclopentane, but the yield was much decreased (entry 3). This detrimental effect on the yield is discussed in the following section. We could employ linker moieties other than diethyl malonate derivatives (entries 4−8). Thus, compounds linked with oxygen or protected nitrogen (6 and 8a−c) gave the corresponding products, substituted tetrahydrofuran and pyrrolidine, in more than 50% yields (entries 5 and 6−8).

The reaction with 4-bromobenzenesulfonamide derivative 8d gave spiro product 9d and $[2 + 2]$ cycloadduct 10 (Scheme 2). Prolonged irradiation severely decreased the yields of both 9d and 10 without improving the relative amount of [9d](#page-2-0). Fortunately, the products 9d and 10 were crystalline compounds and we could obtain single crystals of each that were suitable for X-ray crystallographic analysis. The structures of 9d and 10 with relative stereochemistry were thus unambiguously confirmed (Figure S1, Supporting Information). 2-Methyl- and 2-unsubstituted oxazole derivatives gave no cyclized product with nearly quantit[ative recovery of the](#page-2-0) [start](#page-2-0)ing materials.

Next, we checked the electronic effect of the aryl group at the 2-position of the oxazoles (Table 3). Replacement of the phenyl group with a 4-fluorophenyl group afforded 2f in 69% yield (entry 2), whereas the substrat[e w](#page-2-0)ith a 4-methoxyphenyl group 1g gave the product in somewhat lower yield with

^aAll reactions were carried out in a Pyrex reaction vessel for photochemical reaction by internal irradiation at a concentration of $2.6-5.0$ mM. $\rm ^{b}$ Determined by the $\rm ^{1}H$ NMR integral ratio using 1,1,2,2-tetrachloroethane or pyrazine as an internal standard. The numbers in parentheses are the isolated yields. "Moc = Methoxycarbonyl.

complex byproducts (entry 3). Thus, the electronic property of the substituent on the benzene ring has little effect on the yield or the stereoselectivity of the products.

The spirocyclic compound 2a was stable under the photochemical reaction, as irradiation of isolated 2a caused no decomposition. In contrast, when isolated 3 (containing

Scheme 2. Photoirradiation of 4-Bromobenzenesulfonamide Derivative 8d

Table 3. Electronic Effect of the Substituent at the 2-Position of the Oxazole Ring^a

^aAll reactions were carried out in a Pyrex reaction vessel for photochemical reaction by internal irradiation at a concentration of $2.5-2.7$ mM. \rm^{b} Determined by the $\rm^{1}H$ NMR integral ratio using 1,1,2,2-tetrachloroethane or pyrazine as an internal standard. The numbers in parentheses are the isolated yields.

small amounts (<5%) of unidentified impurities) was irradiated under the same reaction conditions, 2a and 1a were slowly produced to give a mixture consisting of $1a:2a:3 = 5:15:80$ (Scheme 3). These results suggest that the $\begin{bmatrix} 2 & + & 2 \end{bmatrix}$ cycloaddition of 1a forming 3 is reversible and the adduct 3 is not likely an intermediate of the production of 2a.

Scheme 3. Photoirradiation of the $[2 + 2]$ Cycloadduct 3

To obtain insights into the reaction mechanism, we carried out the reaction using a deuterium-labeled compound. When $1a-d_6$ was irradiated under the reaction conditions shown in Table 2, $2a-d_6$ was exclusively produced (Scheme 4). This result clearly shows that the hydrogen at the benzylic position comes [fr](#page-1-0)om the methyl of the prenyl moiety.

When we conducted the reaction using $4-d_3$ ($E/Z = 87:13$), we obtained $5-d_3$ with an H/D ratio of 23:77 (Scheme 5). As

Scheme 4. Reaction with a Hexadeuterated Substrate

Scheme 5. Reaction with a Trideuterated Substrate

shown in Scheme 4, it is clear that the benzylic hydrogen comes only from either of the prenyl methyls. The result shown in Scheme 5 indicates that the benzylic hydrogen mainly comes from the Z-oriented methyl. This observation is consistent with the low yield in the reaction of $1c$ with an (E) -enriched crotyl side chain (Table 2, entry 3).

Though the detailed mechanism is not clear at this stage, we surmise that the p[ath](#page-1-0)way is as shown in Scheme 6 based on the

Scheme 6. Plausible Reaction Pathway

results of Schemes 3, 4, and 5. Photoexcited A would form a biradical intermediate C via the formation of a carbon−carbon bond between the 5-position of oxazole and the alkene moiety, followed by 1,6-hydrogen migration, 18 to give the cyclized product D. As described before, the starting material A reversibly formed the $\lceil 2 + 2 \rceil$ adduct **[B](#page-3-0)**. Direct formation of D from B should be unlikely, as this process would require an intramolecularly impossible hydrogen transfer, taking into account the site selectivity shown in Scheme 5. The overall process can be called a photoene reaction.^{19,20}

In conclusion, we have developed a new method for the photochemical construction of unique spi[rocy](#page-3-0)clic frameworks by the irradiation of oxazole derivatives with an alkene moiety. With appropriate substrates, the reaction proceeds cleanly in a couple of hours to afford the products in synthetically acceptable yields. More importantly, the products are obtained in a highly stereoselective manner in all cases.

■ ASSOCIATED CONTENT

6 Supporting Information

Experimental procedures and spectral data. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATION

Corresponding Authors

*E-mail: n-arai@eng.hokudai.ac.jp. *E-mail: ohkuma@eng.hokudai.ac.jp.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

The authors acknowledge Dr. Tomohiro Seki and Prof. Dr. Hajime Ito (Hokkaido University) for their help in the X-ray crystallographic analysis. This work was supported by JSPS KAKENHI Grant Number 25410105 and the MEXT (Japan) program "Strategic Molecular and Materials Chemistry through Innovative Coupling Reactions" of Hokkaido University.

B REFERENCES

(1) (a) Handbook of Synthetic Photochemistry; Albini, A., Fagnoni, M., Eds.; Wiley-VCH: Weinheim, 2010. (b) Modern Molecular Photochemistry of Organic Molecules; Turro, N. J., Ramamurthy, V., Scaiano, J. C., Eds.; University Science Books: Sausalito, CA, 2010. (c) Fleming, S. A.; Bradford, C. L.; Gao, J. J. Organic Photochemistry; Ramamurthy, V., Schanze, K. S., Eds.; Marcel Dekker, Inc.: New York, 1997; pp 187−243.

(2) (a) Bach, T.; Hehn, J. P. Angew. Chem., Int. Ed. 2011, 50, 1000− 1045. (b) Hoffmann, N. Chem. Rev. 2008, 108, 1052−1103.

(3) Crimmins, M. T.; Reinhold, T. L. Org. React. 1993, 44, 297−588. (4) Hoffmann, N. Synthesis 2004, 481−495.

(5) (a) Zimmerman, H. E.; Armesto, D. Chem. Rev. 1996, 96, 3065− 3112. (b) Armesto, D.; Ortiz, M. J.; Agarrabeitia, A. R. CRC Handbook of Organic Photochemistry and Photobiology, 2nd ed.; Horspool, W. M., Lenci, F., Eds.; CRC Press: Boca Raton, FL, 2004; pp 95/1−95/16. (6) Cycloaddition of furans: (a) Navarro, R.; Reisman, S. E. Org. Lett. 2012, 14, 4354−4357. (b) de la Torre, M. C.; Garcia, I.; Sierra, M. A. J. Org. Chem. 2003, 68, 6611−6618. (c) Crimmins, M. T.; Pace, J. M.; Nantermet, P. G.; Kim-Meade, A. S.; Thomas, J. B.; Watterson, S. H.; Wagman, A. S. J. Am. Chem. Soc. 1999, 121, 10249−10250. (d) Sakamoto, M.; Kinbara, A.; Yagi, T.; Takahashi, M.; Yamaguchi, K.; Mino, T.; Watanabe, S.; Fujita, T. J. Chem. Soc., Perkin Trans. 1 1999, 171−177. (e) Gilbert, A.; Rodwell, P. W. J. Chem. Soc., Perkin Trans. 1 1990, 931−935. (f) Abe, M.; Torii, E.; Nojima, M. J. Org. Chem. 2000, 65, 3426−3431. (g) Griesbeck, A. G.; Stadtmüller, S. Chem. Ber. 1990, 123, 357−362.

(7) Photoreaction of thiophenes: (a) Lablache-Combier, A. Chem. Heterocycl. Compd. 1985, 44, 745−769. (b) D'Auria, M. Gazz. Chim. Ital. 1989, 119, 419−433. (c) Wamhoff, H.; Hupe, H.-J. Tetrahedron Lett. 1978, 125−128. (d) Kobayashi, Y.; Kumadaki, I.; Ohsawa, A.; Sekine, Y.; Ando, A. Heterocycles 1977, 6, 1587−1592. (e) Cantrell, T. S. J. Org. Chem. 1974, 39, 2242−2246. (f) Schenck, G. O.; Hartmann, W.; Steinmetz, R. Chem. Ber. 1963, 96, 498−508.

(8) Photoreaction of thiazoles: (a) D'Auria, M. Targets Heterocycl. Syst. 1998, 2, 233−279. (b) Pavlik, J. W.; Tongcharoensirikul, P.; Bird, N. P.; Day, A. C.; Barltrop, J. A. J. Am. Chem. Soc. 1994, 116, 2292− 2300. (c) Maeda, M.; Kojima, M. J. Chem. Soc., Perkin Trans. 1 1978, 685−692. (d) Saito, I.; Morii, T.; Okumura, Y.; Mori, S.; Yamaguchi, K.; Matsuura, T. Tetrahedron Lett. 1986, 27, 6385−6388. (e) Sindler-Kulyk, M.; Neckers, D. C. Tetrahedron Lett. 1981, 22, 2081−2084.

(9) Photoreaction of oxazoles: (a) Bondock, S.; Griesbeck, A. G. Monatsh. Chem. 2006, 137, 765−777. (b) Griesbeck, A. G.; Bondock, S.; Lex, J. J. Org. Chem. 2003, 68, 9899−9906. (c) Wang, L.; Huang, Y.-C.; Liu, Y.; Fun, H.-K.; Zhang, Y.; Xu, J.-H. J. Org. Chem. 2010, 75, 7757−7768. (d) Huang, C.-m.; Jiang, H.; Wang, R.-z.; Quah, C. K.; Fun, H.-K.; Zhang, Y. Org. Biomol. Chem. 2013, 11, 5023−5033. (e) Šagud, I.; Faraguna, F.; Marinić, Ž.; Šindler-Kulyk, M. J. Org. Chem. 2011, 76, 2904−2908. (f) Lvov, A. G.; Shirinian, V. Z.; Kachala, V. V.; Kavun, A. M.; Zavarzin, I. V.; Krayushkin, M. M. Org. Lett. 2014, 16, 4532−4535.

(10) Photoreaction of isoxazoles: (a) Singh, B.; Ullman, E. F. J. Am. Chem. Soc. 1967, 89, 6911−6916. (b) Griesbeck, A. G.; Franke, M.; Neudörfl, J.; Kotaka, H. *Beilstein J. Org. Chem.* **2011**, 7, 127–134. (c) Pusch, S.; Opatz, T. Org. Lett. 2014, 16, 5430−5433.

(11) Indoles and pyrroles: (a) Weedon, A. In Advances in Photochemistry, Vol. 22; Neckers, D. C., Volman, D. H., von Bünau, G., Eds.; John Wiley & Sons, Inc.: New York, 1997; pp 229−277. (b) González-Béjar, M.; Stiriba, S.-E.; Miranda, M. A.; Pérez-Prieto, J. Org. Lett. 2007, 9, 453−456. (c) Weedon, A. C.; Zhang, B. Synthesis 1992, 95−100. (d) Winkler, J. D.; Scott, R. D.; Williard, P. G. J. Am. Chem. Soc. 1990, 112, 8971−8975. (e) Elliott, L. D.; Berry, M.; Orr-Ewing, A. J.; Booker-Milburn, K. I. J. Am. Chem. Soc. 2007, 129, 3078− 3079. (f) Maskill, K. G.; Knowles, J. P.; Elliott, L. D.; Alder, R. W.; Booker-Milburn, K. I. Angew. Chem., Int. Ed. 2013, 52, 1499−1502.

(12) Arai, N.; Tanaka, K.; Ohkuma, T. Tetrahedron Lett. 2010, 51, 1273−1275.

(13) Arai, N.; Tanaka, K.; Ohkuma, T. Org. Lett. 2012, 14, 1488− 1491.

(14) Thermal [4 + 2] cycloaddition of activated oxazoles are reported: (a) Dean, A.; Ferlin, M. G.; Brun, P.; Castagliuolo, I.; Badocco, D.; Pastore, P.; Venzo, A.; Bombi, G. G.; Di Marco, V. B. Dalton Trans. 2008, 1689−1697. (b) Chan-Huot, M.; Niether, C.; Sharif, S.; Tolstoy, P. M.; Toney, M. D.; Limbach, H.-H. J. Mol. Struct. 2010, 976, 282−289. (c) Nesi, R.; Turchi, S.; Giomi, D.; Papaleo, S. J. Chem. Soc., Chem. Commun. 1993, 978−979.

(15) Photochemical interconversion between regioisomers of phenyloxazole was reported. Maeda, M.; Kojima, M. J. Chem. Soc., Chem. Commun. 1973, 539−540.

(16) MTBE is known as a safer alternative to Et_2O . (a) Little, C. J.; Dale, A. D.; Whatley, J. A.; Wickings, J. A. J. Chromatogr. 1979, 169, 381−385. (b) Tuulmets, A.; Sassian, M. J. Organomet. Chem. 1999, 586, 145−149.

(17) It was reported that product selectivity in some photochemical reactions was improved by conducting the reaction in benzotrifluoride. Dressel, M.; Bach, T. Org. Lett. 2006, 8, 3145−3147.

(18) Similar examples of intramolecular radicalic hydrogen transfer. (a) Wölfle, I.; Chan, S.; Schuster, G. B. J. Org. Chem. 1991, 56, 7313− 7319. (b) Ng, D.; Yang, Z.; Garcia-Garibay, M. A. Tetrahedron Lett. 2002, 43, 7063−7066.

(19) Photochemical ene reactions: (a) Schell, F. M.; Cook, P. M. J. Org. Chem. 1978, 43, 4420−4423. (b) Semin, D. J.; Winkler, P. C.; Rowlen, K. L. Photochem. Photobiol. 1994, 60, 185−195.

(20) A possibility of a photochemically allowed concerted $[4\pi + 4\pi]$ process cannot be ruled out, although it requires high entropy of activation. (a) Ohkura, K.; Nishijima, K.-i.; Seki, K.-i. Photochem. Photobiol. 2001, 74, 385−390. (b) Ohkura, K.; Nishijima, K.-i.; Sakushima, A.; Seki, K.-i. Heterocycles 2000, 53, 1247−1250.